

## THE PHYSICO-CHEMICAL CHARACTERISATION OF SUN FLOWER HONEY FROM BIHOR COUNTY

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### **Abstract**

*Honey has two sources, floral nectar and honeydew. In respect to the floral source, some monofloral honeys are available in every country. Sunflower honey is among them for Romania. In this experiment five samples of Sunflower honey from beekeepers in Bihor County, years 2013, 2014 and 2015 were analysed in order to determinate their physicochemical characteristic: moisture, electrical conductivity, ash, acidity (T, free, lactic, total), sugars (glucose, fructose, sucrose). HMF content and diastase number was also determinate as freshness parameters. All tested parameters comply with both Romanian and European quality requirement and, except for ash content, there are not significant differences to the references values.*

**Keywords:** Sunflower honey, physicochemical characterisation, quality criteria

### **INTRODUCTION**

Honey is an animal origin food produced by bees (*Apis mellifera*) from two different sources, floral nectar and honeydew, which is the excretions of plant sucking insects (*Hemiptera*) on the living part of plants or secretions of living parts of plants (Mărghitaș, 2005). Due to its high sugars content honey is used as a healthy natural sweetener and as an important energetic source (1,272 kJ (304 kcal) /100 g). Fructose and glucose represent minimum 60% of honey sugars (Zielińska et al, 2014), but more than 25 oligosaccharides were identified too (Anklam, 1998) and until sugar were produced on an industrial scale, after 1800, honey was the only sweetener used from millennia. Apart from water which must be under 20%, minor components of honey are antioxidants such as phenolic compounds (phenolic acids and flavonoids), amino acids as proline, vitamins (vitamin C) and enzymes as catalase, peroxidase, glucose-oxidase, amylase. Among those components, fructose which can transform in hydroxymethylfurfural (HMF) (White, 1994) and amylase are related to a specific freshness indicator or to an improper technological handling (Tosi, 2002, Turhan, 2008).

Beside its nutritional value, honey has multiple applications for improving human health (Bogdanov, 2010). A branch of alternative medicine called apitherapy offer a great deal of cures based on honey and others bee's products. The main effects are antimicrobial, antiviral and antiparasitical (Molan 1992 a, b, 1997). On the other hand, the phenolic compounds were found to be responsible for the antioxidant capacity of honey (Utthuty, 2011). Many studies emphasise that antioxidant activity is correlated to the botanical origin of honey and their source (Weselowska et al, 2014, Vela et al, 2007, Baltrusaityte et al., 2007, Al-Mamary, 2002). Honey effect on cancer was studied recently (Orsolic, 2009) and the researches were review by Erejuwa et al, 2014.

In respect to the floral source, some monofloral honeys are available in every country, depending of their climate (Purcărea et al, 2014, Şarba et al, 2014). For Romania, more specific the western part, the most predominant as quantity are acacia honey, rape honey, lime honey but there are in smaller amount other types as: Chestnut, heather, dandelion, fruit trees. Sunflower (*Helianthus Annuus* L. – ASTERACEAE) is cultivated in Bihor County on large areas for its oily seeds, it represents an important source of nectar and pollen to bees so Sun Flower honey can be obtain as monofloral assortment. The variability of the floral source regarding the pollen production, leads to a big variability in pollen content, from 20% to 90%, with a PG/10 g value mostly below 30 000 (Persano Oddo and Piro, 2004).

## MATERIALS AND METHODS

### Materials

The tested materials consist in five samples of Sunflower (*Heliantus Annus.*) honey, two from 2013, two from 2014 and one from 2015. The samples were purchased directly from beekeepers in Bihor County. All samples were purchased in glass bottles and stored in dark at ambient temperature below 25°C until analyzed. The samples were liquefied by gentle warming at 40°C in a thermostatic bath.

All the used reagents were p.a. grade: sodium metabisulphite potassium hexacyanoferrate(II),  $K_4Fe(CN)_6 \cdot 3H_2O$ , double sublimate Iodine and starch from Merck Germany,  $Zn(CH_3.COO)_2 \cdot 2H_2O$ , KCl, NaOH, HCl,  $CuSO_4 \cdot 7H_2O$ , Seignette salt, NaCl,  $NaHCO_3$ ,  $Na_2S_2O_7$  from Chimopar Romania.

The laboratory devices used in the present study were: Vortex Hettrich, Digital pH-meter/Conductometer SensiION HACH, Thermostatic Oven Nitech Pol Eko-Aparatura, Calcination Oven

Nabertherm, LE 2/11/R6, spectrophotometer UVMini-1240 Shimadzu, Digital refractometer Kruss Germany.

### **Methods**

The quality parameters were determined according to the instructions of the International Honey Commission (Bogdanov, 2002) and SR 784-3/ 2009 excepted for glucose content (Popescu et al, 1986) and free, lactic and total acidity (Silva et al, 2009).

3.1. Moisture content - refractometric method. All measurements were corrected for temperature and the corresponding % moisture (g/100 g honey) was obtained from the table for the purpose.

3.2. Electrical conductivity - was determined for a solution corresponding to 20g dry matter of honey in 100 ml distilled water by conductometric assay. The cell constant was determined by measuring the conductivity of a 0,1M potassium chloride solution and the results were corrected for the temperature of the sample.

3.3. Ash content - After the removal of water from the samples on an electric plate, the honey is burnt to ashes at 600°C and the residue is weighed.

3.4 pH - was measured in solution of 10 g honey in 75 ml of CO<sub>2</sub> free distilled water.

3.5 Free, lactic and total acidity - were determined as follows, by titrimetric method: the addition of 0.05 M NaOH was stopped at pH 8.50 (free acidity), immediately a volume of 10 ml 0.05 M NaOH was added and, without delay, back-titrated with 0.05 M HCl to pH 8.30 (lactic acidity). Total acidity results were obtained by adding free and lactic acidities.

3.6 Acidity, °T – 10% honey solution was titrated with NaOH solution 1N until pale pink (phenolphthalein)

3.7 HMF content - spectrophotometric White method. This method involves measurement of UV absorbance of clarified aqueous honey solutions with and without sodium metabisulphite. The absorbance of the solutions at 284 and 336 nm was determined.

3.8 Diastase activity- Shade method using a buffered solution of soluble starch and honey incubated in a thermostatic bath at 40 °C. The diastase value was calculated using the time taken for the absorbance to reach 0.235, and the results were expressed in Gothe degrees as the amount (mL) of 1% starch hydrolyzed by an enzyme in 1 g of honey in 1 h.

3.9 Sugars - Reducing sugars were determined by Elseriodometric method. Glucose was determined by Auerbach and Bodlander method when only sugar whit aldehydic function is reduced by iodine in basic environment. The difference between inverted sugars and glucose gave fructose. The difference in concentrations of inverted sugar before and after

the hydrolysis procedure (inversion) was multiplied by 0.95 to reach the apparent sucrose content.

Procedures were applied on fresh samples and the experiment was performed between July 2013 and July 2015

Statistical analysis -All tests were performed as triplicate and are presented as mean  $\pm$  SD. The results of different experimental variants were compared by T-test for independent samples. Differences between means at 95% ( $p < 0.05$ ) confidence level were considered statistically significant.

## RESULTS AND DISCUSSIONS

The obtained values are shown in Table 1 for the physicochemical characteristics and in Table 2 for the freshness parameters. The results are compared to those accepted by European Union (Council Directive 2001/110/EC, 2001) and Romanian Legislation (SR 784/1,2-2009). In the same time the tables presents the reference values for this type of honey which proceed from the descriptive sheets of the main European monofloral honeys (Persano Oddo and Piro, 2004), a reliable source of a large work from The International Honey Commission and Apimondia (IHC). As for Sunflower honey the number of data taken into consideration was between 157 for sucrose value and 243 for electrical conductivity.

Table 1

Physicochemical characterisation of Sunflower honey (n=5)

Nr.	Parameter	UM	Experimental values Mean $\pm$ SD	Legislation limit		Reference value mean $\pm$ SD
				CE nr. 2001/110	SR 784- 1,2/ 2009	
1	Moisture	g/100 honey g	18,7 $\pm$ 1.18	Max 20	Max 20	17,8 $\pm$ 1,1
2	Electrical conductivity	mS/cm	0.301 $\pm$ 0.059	Min 0,8	Min 0,8	0.34 $\pm$ 0,08
3	Ash content	Mg/kg	0,112 $\pm$ 0.018	-	0.5	-
4	pH	pH units	3.656 $\pm$ 0.185	-	-	3.8 $\pm$ 0.2
5	Acidity	<sup>0</sup> T	2.408 $\pm$ 0.472	-	4	-
6	Free acidity	MEQ/1000 g	22.36 $\pm$ 1.045	Max 50	-	23.1 $\pm$ 6.3
7	Lactones	MEQ/1000 g	9.892 $\pm$ 0.247	-	-	10.1 $\pm$ 5.8
8	Total acidity	MEQ/1000 g	32.252 $\pm$ 1.24	-	-	32.1 $\pm$ 5.8
9	Fructose + Glucose	g/100 g	76.54 $\pm$ 2.731	Min 60	Min 70	76.7 $\pm$ 2.7
10	Glucose	g/100 g	37.88 $\pm$ 1.37	-	-	37.4 $\pm$ 1.5
11	Fructose	g/100 g	38.66 $\pm$ 1.472	-	-	39.2 $\pm$ 1.6
12	Fructose/Glucose		1,021 $\pm$ 0.021	-	-	1,05 $\pm$ 0.04
13	Sucrose	g/100 g	0.528 $\pm$ 0.301	Max 5	Max 5	0.3 $\pm$ 0.2
14	Glucose/Water		2.05 $\pm$ 0.145	-	-	2.1 $\pm$ 0,13

From both Romanian and European regulations (parameters 1, 2, 3, 5, 9, 13 from Table 1) the tested samples show a good frame to the admitted values. Regarding to the reference values, the statistic analysis (Student test) shows no significant differences ( $p>0.05$ ) between all obtained experimental values and the reference ones excepted the sucrose content. In this case, significant differences occur between the observed mean values. The experimental ratios involving sugars content (F/G and G/W) comply with the references values and are typical for this kind of honey (PersanoOddo and Piro, 2004).

The experimental values for water content, pH, electrical conductivity, ash, sugars (glucose, fructose and sucrose) the investigated parameters comply to reported ones in Romania (Isopecu et al, 2014, Stihi et al, 2013, Diacu et Tănțăveanu, 2007, Mărghitaș et al, 2008) and in other countries like Turkey - Sari and Ayyeldiz, 2012 and Tukac et al, 2007 and France - Devilliers and al, 2004.

Table 2

Freshness parameters of Sunflower honey (n=5)

Nr.	Parameter	UM	Experimental values Mean $\pm$ SD	Legislation limit		Reference value mean $\pm$ SD
				CE nr. 2001/110/	CE nr. 2001/110/	
1	HMF	mg/kg	0.71 $\pm$ 0.593	Max 40	Max 10/15*	-
2	Diastase activity	Shade scale	15.01 $\pm$ 0.903	Min 8	Min 10.9	20.8 $\pm$ 5.6

\*The values refer to the artisanal, respectively to the marketable honey

As for the freshness parameters,(Table 2), experimental values are much under (for HMF) of above the limitation even if Romanian requirements are less permissive than the European Directive. The value for HMF content really demonstrates the freshness of the tested samples and because HMF practically doesn't exists in fresh honey (White, 1994). In the same time it demonstrate the lack of thermal mistreatment often applied on Sunflower honey due to its quickly crystallisation rate. Isopescu et al, 2014 find the same situation for Sunflower honey collected in 2011-2012. Higher values reported by Devilliers and al, 2004(3.15 $\pm$ 1.85) and Tukac et al, 2007 (2,3 $\pm$ 1.43) are acceptable but demonstrate the necessity of thermal treatment for packaging.

## CONCLUSIONS

The experiment leads to some conclusions, as follows:

The Sunflower honey samples tested meet both Romanian and European minimal quality criteria for this type of honey. The HMF content and Diastase number certify the freshness of the product and seem to indicate that no thermal treatment was performed and the samples are really from the harvesting year declared by beekeepers.

Except the ash content, the tested samples have similar physicochemical properties as the reference ones for this type of honey in Europe. The values for ash were not an accident because the samples came from the same area but in different years. Further determination on mineral content could give more clarifications in this subject as well as the extension of the experiment on a greater number of samples.

Further testing on greater number of samples, with different initial HMF values are needed in order to extrapolate the mathematical correlation between HMF content and periods.

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